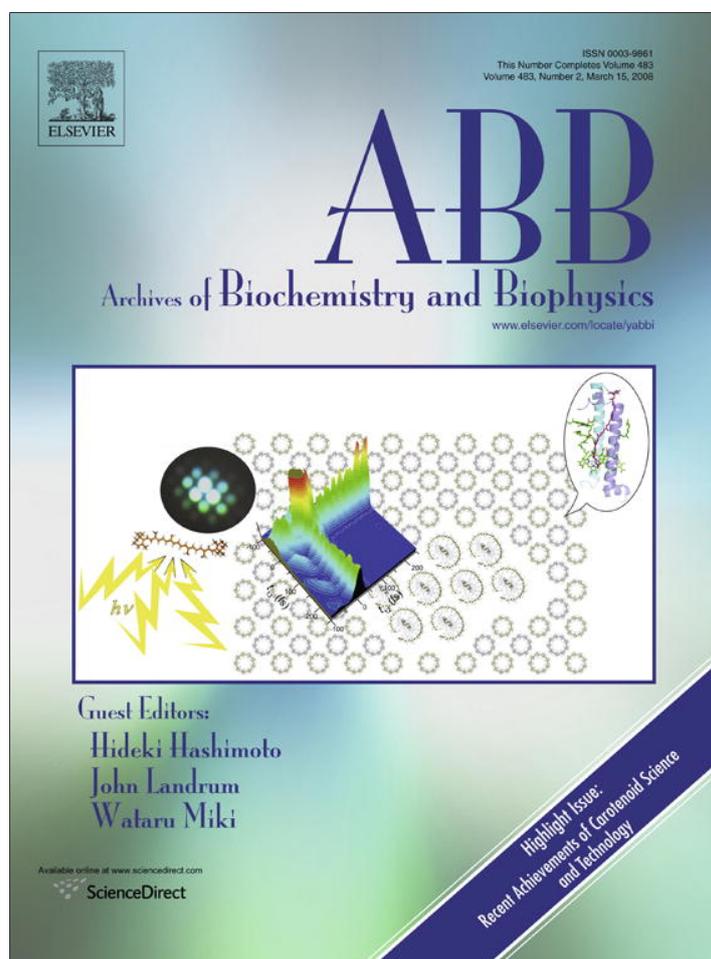


Provided for non-commercial research and education use.
Not for reproduction, distribution or commercial use.



This article appeared in a journal published by Elsevier. The attached copy is furnished to the author for internal non-commercial research and education use, including for instruction at the authors institution and sharing with colleagues.

Other uses, including reproduction and distribution, or selling or licensing copies, or posting to personal, institutional or third party websites are prohibited.

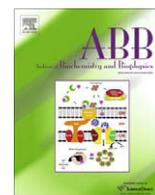
In most cases authors are permitted to post their version of the article (e.g. in Word or Tex form) to their personal website or institutional repository. Authors requiring further information regarding Elsevier's archiving and manuscript policies are encouraged to visit:

<http://www.elsevier.com/copyright>



Contents lists available at ScienceDirect

Archives of Biochemistry and Biophysics

journal homepage: www.elsevier.com/locate/yabbi

Review

Recent progress in structural studies of carotenoids in animals and plants

Takashi Maoka*

Research Institute for Production Development, Food Function and Chemistry, 15 Shimogamo-morimoto-cho, Sakyo-ku, Kyoto 606-0805, Japan

ARTICLE INFO

Article history:

Received 28 August 2008
and in revised form 7 October 2008
Available online 1 November 2008

Keywords:

Carotenoids
New structures
Structural elucidation
Aquatic animals
Higher plants

ABSTRACT

About 750 naturally occurring carotenoids had been reported as of 2004. Furthermore, annually, more than 20 new structures of carotenoids are reported. Improvements of analytical instruments such as NMR, MS, HPLC, etc., have made it possible to perform the structural elucidation of very minor carotenoids in nature. Interesting new structural carotenoids can still be identified in aquatic animals and higher plants. The present paper provides a review of new structural carotenoids isolated from aquatic animals and higher plants by our group over the last five years.

© 2008 Elsevier Inc. All rights reserved.

Since the first structural elucidation of β -carotene by Kuhn and Karrer in 1928–1930, about 750 naturally occurring carotenoids had been reported as of 2004 [1]. Improvements of analytical instruments such as NMR¹, MS, HPLC, etc., have made it possible to perform the structural elucidation of very minor carotenoids in nature [2,3]. Annually, more than 20 new structures of carotenoids are being reported.

Our research group has developed structural analysis of naturally occurring carotenoids [3] using NMR, MS, MS/MS [4], LC/MS [5], etc., over the last decade. Herein, I describe the progress in structural studies of naturally occurring carotenoids of our group.

Carotenoids in aquatic animals

Marine animals, especially bivalves (oysters, clams, scallops, etc.) contain various carotenoids, which show structural diversity [6–9]. Bivalves accumulate carotenoids from their dietary algae and modify them through metabolic reactions. Many of the carotenoids present in bivalves are metabolites of fucoxanthin, peridinin, and diatoxanthin [6–9].

In 1992, our group isolated two unique carotenoids, named crassestreaxanthin A (**1**) and B (**2**), from the Japanese oyster *Crassostrea gigas* [10]. These two carotenoids have unique structural end groups, which have not been previously reported. These struc-

tures were determined to be 3',6'-epoxy-3-hydroxy-6'-methyl-7,8-didehydro-1',2',5',6',7',8'-hexahydro-16'-nor- β , ψ -carotene-1',8'-dione (**1**) and 3,3'-dihydroxy-7,8-didehydro-1',2',7',8'-tetrahydro-6'-methyl-16'-nor- β , ψ -carotene-1',8'-dione (**2**), respectively, by the detailed analysis of MS and NMR data [10] as shown in Scheme 1. Based on the structural similarity, crassestreaxanthin A and B were assumed to be metabolites of halocynthiaxanthin. The (3R, 3'S) configurations of crassestreaxanthin B were confirmed based on total synthesis by Tode et al. [11]. Further studies of carotenoids in marine animals revealed that crassestreaxanthin A and B and their 3-acetates were widely distributed in marine bivalves. Moreover, two new carotenoids, **3** and **4**, possessing a structural moiety of crassestreaxanthin A, were isolated from the oyster as minor components [12,13].

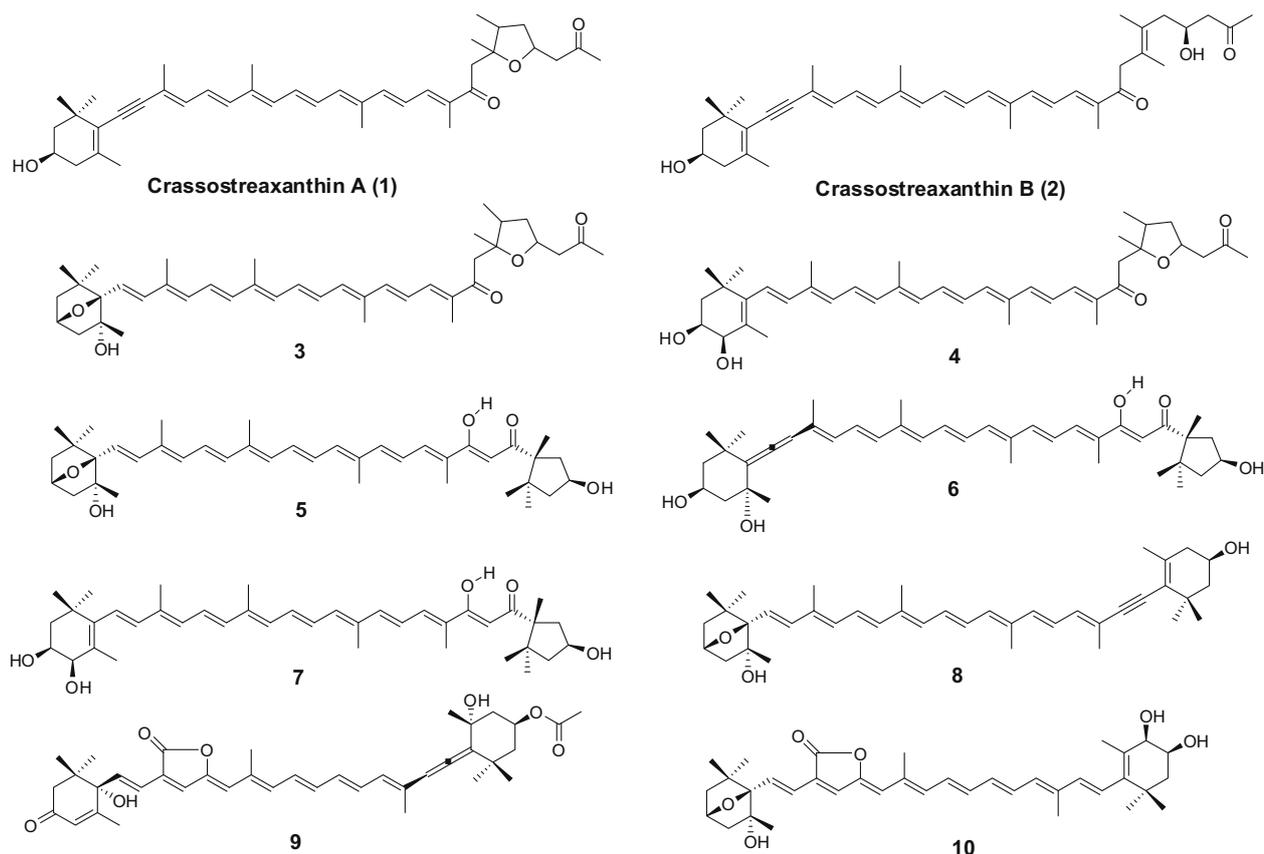
Mytiloxanthin, which has a unique enol hydroxy group at C-8' in the polyene chain and a 3'-hydroxy-6'-oxo- κ -end group, is a characteristic carotenoid in marine mussels and oysters [9]. Furthermore, three mytiloxanthin analogs containing a 3,6-epoxy-end group (**5**), an allenic end group (**6**), and a 3,4-dihydroxy- β -end group (**7**) were isolated from the oyster [12,13]. Compound **6** was assumed to be a metabolic intermediate from fucoxanthin to mytiloxanthin. A 3,6-epoxy derivative of diadinoxanthin, named cycloidadinoxanthin (**8**), was also isolated from the oyster [12]. In addition to these C₄₀-skeletal carotenoids described above, two new C₃₇-skeletal carotenoids, **9** and **10**, were isolated from the oyster [12,13]. Compound **9**, having a 6-hydroxy-3-oxo- ϵ -end group, was assumed to be a metabolite of peridinin (Scheme 1).

Some edible clams have a bright orange or red color due to the presence of carotenoids. Fucoxanthin 3-ester (**11**) and fucoxanthinol 3-ester (**12**) were found to be major carotenoids in *Macrura chinensis* [14], *Ruditapes philippinarum*, and *Meretrix petechialis*.

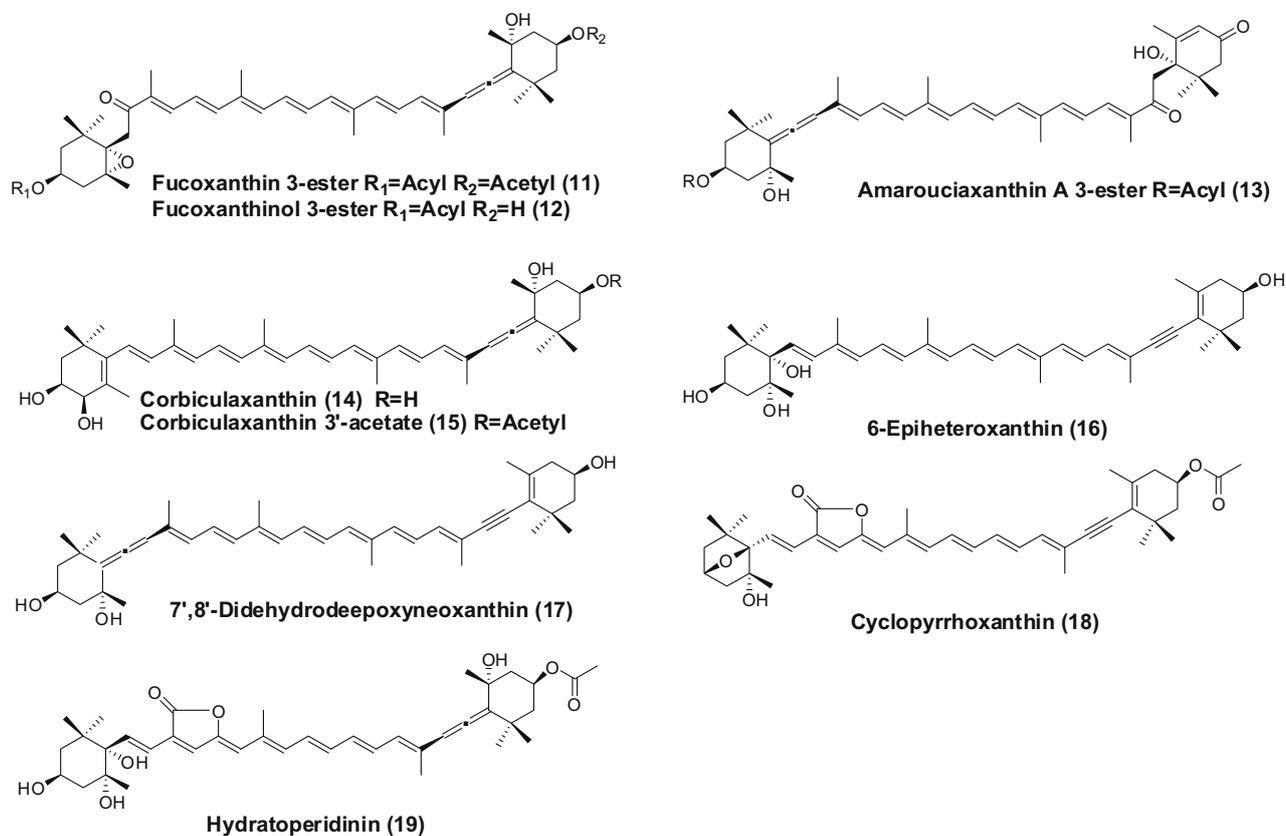
* Fax: +81 75 781 1118.

E-mail address: maoka@mbox.kyoto-inet.or.jp.

¹ Abbreviations used: NMR, nuclear magnetic resonance; MS, mass spectrometry; HPLC, high performance liquid chromatography; MS/MS, mass spectrometry/mass spectrometry; LC/MS, Liquid chromatography mass spectrometry; CD circular dichroism spectrometry, LC/NMR Liquid chromatography nuclear magnetic resonance.



Scheme 1.



Scheme 2.

Amarouciaxanthin A 3-ester (**13**) was also identified as a major carotenoid in *Paphia amabilis*. These structures were characterized using ^1H NMR and MS spectrometry. Fatty acids esterified with these carotenoids were assigned as C24:6, C22:5, C22:6, C20:5, C20:0, C20:1, C18:0, C18:1, C16:0, C16:1, and C14:0 based on FAB-MS data.

There are many reports on carotenoids in marine shellfish [6–9]. However, there are few reports on the carotenoids of shellfish inhabiting brackish or fresh water [6–9]. In Japan, *Corbicula japonica* inhabits brackish water and *C. sandai* and *C. leana* inhabit fresh water. Therefore, we studied the carotenoids of these three species of *Corbicula* clams.

Forty-three carotenoids, including six new ones, were isolated from the three species of *Corbicula* clams [15,16]. Among them, peridinin derivatives were found only in *C. japonica*, which inhabits brackish water. On the other hand, lutein and loroxanthin were found only in *C. sandai* and *C. leana*, which inhabit fresh water. Fucoxanthin and diatoxanthin derivatives were found in all three species [15]. From the carotenoid patterns in *Corbicula* clams, it can be concluded that the brackish water clam mainly feeds on dinoflagellates and diatoms and freshwater clams mainly feed on green algae and diatoms.

Four new carotenoids, corbiculaxanthin (**14**), corbiculaxanthin 3'-acetate (**15**), 6-epiheteroxanthin (**16**), and 7',8'-didehydrodeoxyneoxanthin (**17**), were found in all the above three species of clam but not in other clam species [15,16]. Therefore, they are assumed to be carotenoids peculiar to the genus *Corbicula*. 7',8'-Didehydrodeoxyneoxanthin (**17**) has an interesting structure with both allenic and acetylenic moieties. Two new metabolites of peridinin, cyclopyrrhoxanthin (**18**), and hydratoperidinin (**19**), were isolated from *C. japonica*, which inhabits brackish water

[15,16] (Scheme 2). Carotenoids found in these animals provide a key to the food chain as well as metabolic pathways [9].

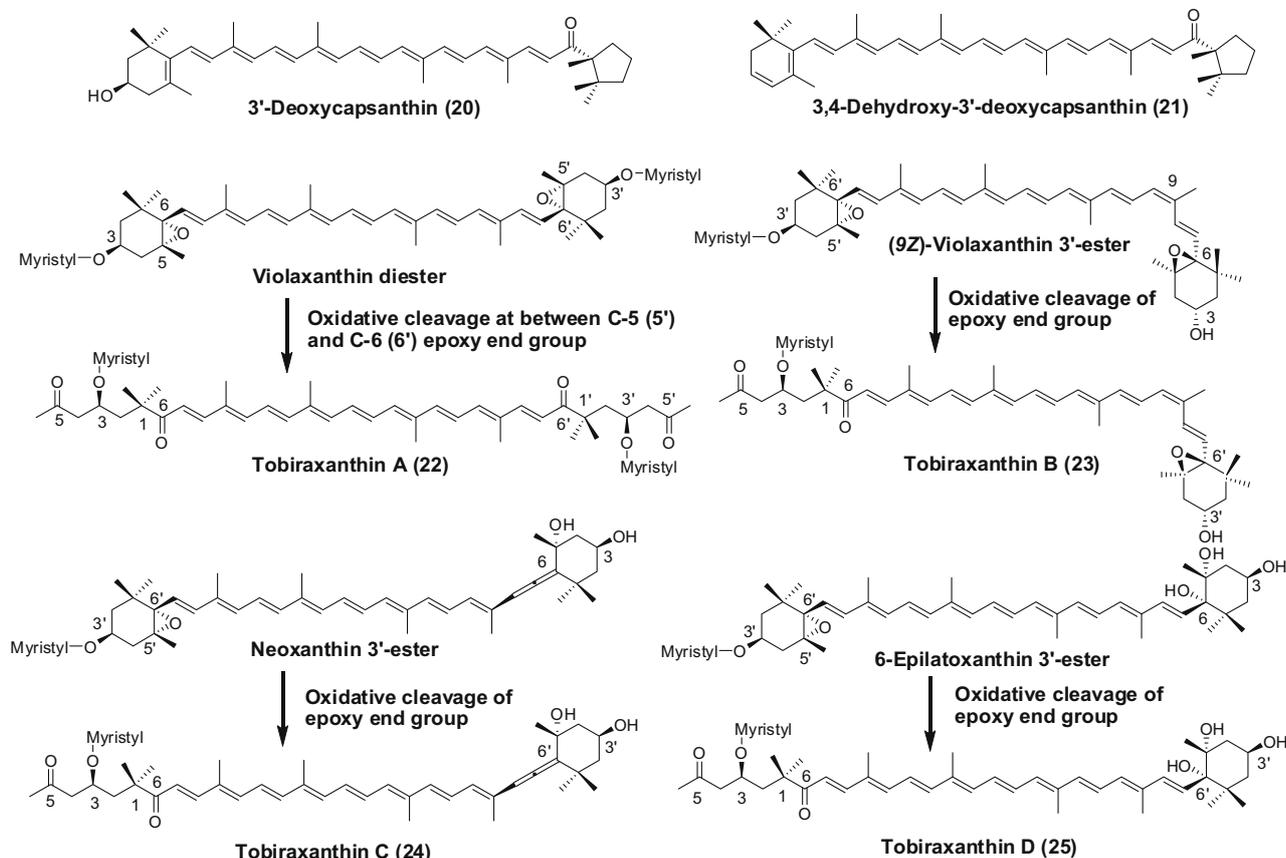
Carotenoids in higher plants

Carotenoids present in photosynthetic organs have been extensively investigated. These biosynthetic roots were revealed by enzymatic and gene level experiments [17]. On the other hand, carotenoids present in non-photosynthetic organs such as fruits, seeds, and flowers, showed structural diversity. Some of the carotenoids found in non-photosynthetic organs were formed by secondary metabolic reactions, such as oxidation, the cleavage of polyene chains, (*Z/E*) (*cis-trans*) isomerization, etc. [17]. Therefore, new structural carotenoids can still be found in fruits, seeds, and flowers in higher plants.

Carotenoids with 6-oxo- κ -end group in paprika

Paprika, *Capsicum annum*, is a good source of carotenoids. Major carotenoids in paprika are capsanthin and capsorubin, containing a 3-hydroxy-6-oxo- κ -end group. These carotenoids show a strong singlet oxygen quenching activity [18] and also the inhibition of lipid peroxidation induced by free radicals [19]. They also possess excellent anti-cancer activities [20].

Several unique structural carotenoids have been reported in paprika by the University of Pécs group in Hungary [21]. Our group also isolated a series of apocarotenoids derived from capsanthin from paprika [22]. Furthermore, two new carotenoids, named 3'-deoxycapsanthin (**20**) and 3,4-dehydroxy-3'-deoxycapsanthin (**21**), were isolated from paprika as very minor components [23]. These carotenoids have a 6-oxo- κ -end group. Carotenoids with a

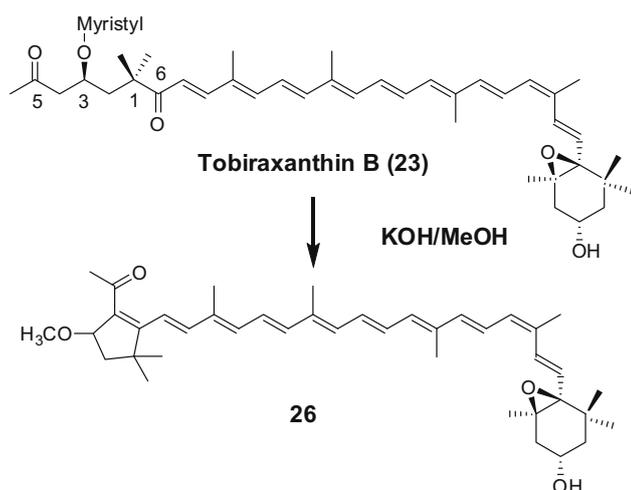


Scheme 3.

6-oxo- κ -end group have not been reported previously in nature. These carotenoids were assumed to be formed from β -cryptoxanthin 5',6'-epoxide by pinacolic rearrangement through capsanthin capsorubin synthase [23].

Seco-carotenoids in the seeds of Pittosporum tobira

Pittosporum tobira is a small, slender, evergreen tree growing in southern Japan. In summer, the seeds have a pale yellow color. In autumn, the seeds are exposed to sunlight and change color from yellow to red.



Reaction of alkaline medium with tobiraxanthin B

Scheme 4.

The major carotenoids in the yellow seeds are violaxanthin, (9Z)-violaxanthin, antheraxanthin, and neoxanthin esters. On the other hand, a series of acylated seco-carotenoids having a 3-acyloxy-5,6-diseco-5,6-diketo- β -end group, named tobiraxanthin A (**22**), B (**23**), C (**24**), and D (**25**), were isolated from red seeds as major carotenoids [24]. These structures were elucidated by the detailed analysis of MS and NMR spectral data. Tobiraxanthin A, B, C, and D are corresponding oxidative cleavage products of C5–C6 (C5'–C6') bond(s) in the violaxanthin diester, (9Z)-violaxanthin 3'-ester, neoxanthin 3'-ester, and 6-epilatoxanthin 3'-ester, respectively as shown in Scheme 3.

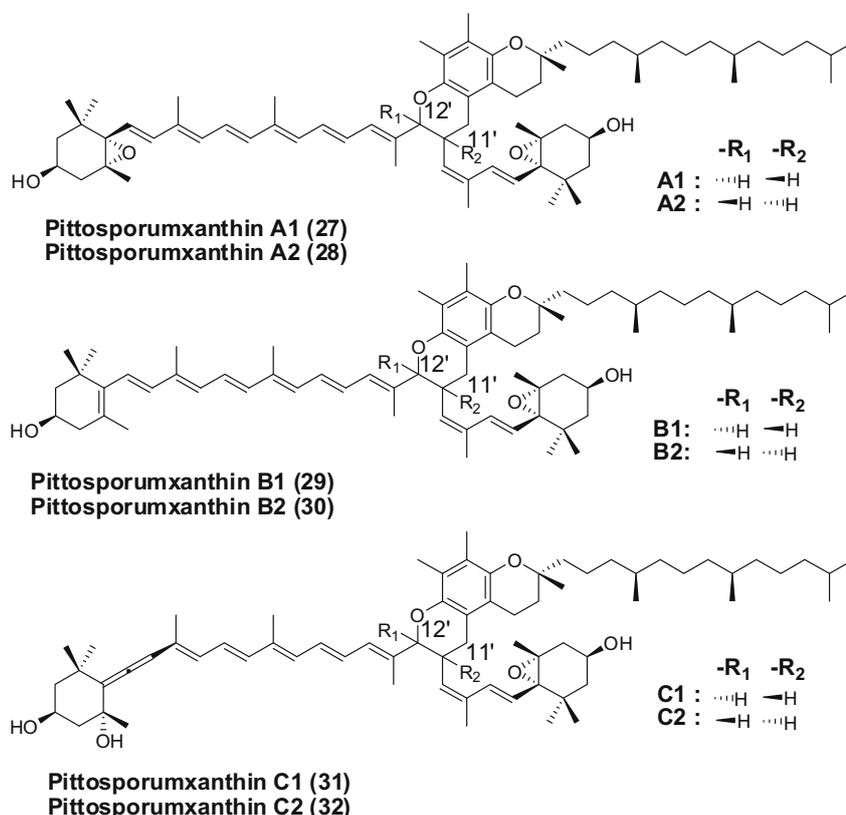
These seco-carotenoids (**22–25**) showed on approximately 15–30 nm longer wavelength shift than their corresponding precursor epoxy carotenoids because of the introduction of conjugated carbonyl group(s) at C6 (C6') by the oxidative cleavage of C5–C6 (C5'–C6') bond(s). Therefore, it was revealed that the color change from yellow to red in the seeds is due to the formation of these seco-carotenoids from epoxy carotenoids. Furthermore, it was assumed that these seco-carotenoids protected the seeds from oxidation induced by sunlight and singlet oxygen because it was reported that the introduction of the conjugated double bond system in carotenoids enhanced the singlet oxygen quenching activity [25,26].

Moreover, seco-carotenoids with a 3-acyloxy-5-hydroxy-5,6-diseco-6-keto- β -end group were also isolated from the red seeds [27].

These seco-carotenoids showed lability in alkaline medium. Tobiraxanthin B was converted to a carotenoid with a 3-methoxy-5-keto-5,6-seco-4,6-cyclo- β -end group (**26**) on treatment with KOH/MeOH [24] as shown in Scheme 4.

Carotenoid and α -tocopherol complexes in the seeds of P. tobira

A series of carotenoid and α -tocopherol complexes named pittosporumxanthins were isolated from the red-colored seeds of *P.*



Scheme 5.

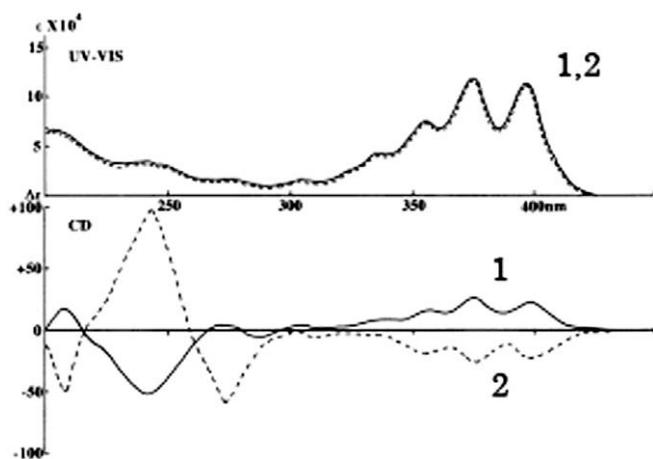


Fig. 1. UV-vis (top) and CD (bottom) spectra of **1**: pitto- and pittedoxanthin A1 and **2**: pitto- and pittedoxanthin A2 in ether.

tobira [28,29]. Pitto- and pittedoxanthin A1 (**27**) and A2 (**28**) were diastereomeric pairs of the cycloaddition product of (9*Z*)-violaxanthin at the C-11' and C-12' positions, with α -tocopherol showing a O-C12'-C11'-C28'' linkage. Furthermore, (9*Z*)-antheraxanthin and α -tocopherol complexes, named pitto- and pittedoxanthin B1 (**29**) and B2 (**30**) and (9*Z*)-neoxanthin, and α -tocopherol complexes, named pitto- and pittedoxanthin C1 (**31**) and C2 (**32**), were isolated. These structures were elucidated by the detailed analysis of NMR and MS, including MS/MS experiment and CD spectral data as shown in Scheme 5. The absolute configurations of the carotenoid end groups in the pitto- and pittedoxanthins were determined by a modified Mosher's method using ^1H NMR spectroscopy. Through the addition of α -tocopherol at the C-11' and C-12' positions in the polyene chain of carotenoids, two diastereomeric pairs were formed. Mirror image CD spectra were observed between diastereomeric pairs of pitto- and pittedoxanthins A1 and A2, between B1 and B2, and between C1 and C2. Fig. 1 shows the UV-vis and CD spectrum of pitto- and pittedoxanthins A1 and A2. These CD spectra were assumed to have originated from excitation coupling of $\pi \rightarrow \pi^*$ absorption of hexaene (or heptaene) around 370 nm and diene around 240 nm. Therefore, from the excitation chirality rule of the CD spectrum [30], the (11'*R*, 12'*S*) chirality for pitto- and pittedoxanthins A1, B1, and C1 and (11'*S*, 12'*R*) chirality for pitto- and pittedoxanthins A2, B2, and C2 were determined.

Pitto- and pittedoxanthins are reaction products of epoxy carotenoids and α -tocopherol. Therefore, it was assumed that epoxy carotenoids such as violaxanthin, antheraxanthin, and neoxanthin in seeds of *P. tobira* might quench α -tocopherol radical through the formation of adduct products in their polyene chain.

Conclusion

Interesting new structural carotenoids can still be found in plants and animals. The improvement of analytical tools such as

NMR, MS, CD, HPLC, LC/MS, LC/NMR etc., has made it possible to perform the structural elucidation of very minor carotenoids. Structures of new carotenoids provide information on the function, biosynthesis in plants, and a key to the food chain as well as metabolic pathways in animals.

Acknowledgments

I thank Dr. N. Akimoto of Graduate School of Pharmaceutical Sciences, Kyoto University and collaborators at Kyoto Pharmaceutical University for supporting this work.

References

- [1] G. Britton, S. Liaaen-Jensen, H. Pfander (Eds.), Carotenoids Hand Book, Birkhäuser, Basel, Switzerland, 2004.
- [2] G. Britton, S. Liaaen-Jensen, H. Pfander (Eds.), Carotenoids Volume 1B: Spectroscopy, Birkhäuser, Basel, Switzerland, 1995.
- [3] T. Maoka, N. Akimoto, Carotenoid Science 13, (2008).
- [4] T. Maoka, Y. Fujiwara, K. Hashimoto, N. Akimoto, Lipids 39 (2004) 179–183.
- [5] T. Maoka, Y. Fujiwara, K. Hashimoto, N. Akimoto, J. Oleo, Sciences 51 (2002) 1–9.
- [6] S. Liaaen-Jensen, Pure Appl. Chem. 63 (1991) 1–12.
- [7] A.Z. Mercadante, Pure Appl. Chem. 71 (1999) 2263–2722.
- [8] T. Matsuno, Fisheries Sci. 67 (2001) 771–789.
- [9] S. Liaaen-Jensen, Carotenoids in food chain, in: G. Britton, S. Liaaen-Jensen, H. Pfander (Eds.), Carotenoids Volume 3: Biosynthesis and Metabolism, Birkhäuser, Basel, Switzerland, 1998, pp. 359–371.
- [10] Y. Fujiwara, T. Maoka, Tetrahedron Lett. 42 (2001) 266–2693.
- [11] C. Tode, Y. Yamano, M. Ito, M. J. Chem. Soc. Perkin Trans. 1 (2001) 3338–3345.
- [12] T. Maoka, K. Hashimoto, N. Akimoto, Y. Fujiwara, J. Nat. Prod. 64 (2001) 578–581.
- [13] T. Maoka, Y. Fujiwara, K. Hashimoto, N. Akimoto, Chem. Pharm. Bull. 53 (2005) 1207–1209.
- [14] T. Maoka, Y. Fujiwara, K. Hashimoto, N. Akimoto, J. Agric. Food Chem. 55 (2007) 1563–1567.
- [15] T. Maoka, Y. Fujiwara, K. Hashimoto, N. Akimoto, J. Agric. Food Chem. 53 (2005) 8357–8364.
- [16] T. Maoka, Y. Fujiwara, K. Hashimoto, N. Akimoto, J. Nat. Prod. 68 (2005) 1341–1344.
- [17] G. Britton, Overview of carotenoid biosynthesis, in: G. Britton, S. Liaaen-Jensen, H. Pfander (Eds.), Carotenoids Volume 3: Biosynthesis and Metabolism, Birkhäuser, Basel, Switzerland, 1998, pp. 13–147.
- [18] H. Matsufuji, H. Nakamura, M. Chino, M. Takeda, J. Agric. Food Chem. 46 (1998) 3468–3472.
- [19] T. Maoka, Y. Goto, K. Isobe, Y. Fujiwara, K. Hashimoto, K. Mochida, J. Oleo. Sci. 50 (2001) 663–665.
- [20] T. Maoka, K. Mochida, M. Kozuka, Y. Ito, Y. Fujiwara, K. Hashimoto, F. Enjo, M. Ogata, Y. Nobukuni, H. Tokuda, H. Nishino, Cancer Lett. 172 (2001) 103–109.
- [21] J. Deli, P. Molnár, Curr. Org. Chem. 6 (2002) 1197–1219.
- [22] T. Maoka, Y. Fujiwara, K. Hashimoto, N. Akimoto, J. Agric. Food Chem. 49 (2001) 1601–1606.
- [23] T. Maoka, N. Akimoto, Y. Fujiwara, K. Hashimoto, J. Natur. Prod. 67 (2004) 115–117.
- [24] Y. Fujiwara, K. Hashimoto, K. Manabe, T. Maoka, Tetrahedron Lett. 43 (2002) 4385–4388.
- [25] O. Hirayama, K. Nakamura, S. Hamada, Y. Kobayashi, Lipids 29 (1994) 149–150.
- [26] N. Shimizu, M. Goto, W. Miki, Fisheries Sci. 62 (1996) 134–137.
- [27] T. Maoka, Y. Fujiwara, K. Hashimoto, N. Akimoto, Phytochemistry 67 (2006) 2120–2125.
- [28] Y. Fujiwara, T. Maoka, Tetrahedron Lett. 42 (2001) 266–2699.
- [29] T. Maoka, N. Akimoto, Y. Kuroda, K. Hashimoto, Y. Fujiwara, J. Nat. Prod. 71 (2008) 622–627.
- [30] N. Harada, K. Nakanishi, Circular Dichroic Spectroscopy–Exciton Coupling in Organic Stereochemistry, University Science Book, 1983.